



education

Department of
Education
FREE STATE PROVINCE

PRACTICAL TASK / PRAKTISE TAAK

GRADE 11 / GRAAD 11

**PHYSICAL SCIENCES
FISIESE WETENSKAPPE**

MEMORANDUM

JUNE 2018 / JUNIE 2018

MARKS: 15 / PUNTE: 15

TIME: 30 MINUTES / TYD: 30 MINUTE

This memorandum consists of THREE pages.
Hierdie memorandum bestaan uit DRIE bladsye.

QUESTION 1 / VRAAG 1

1.1 Marking criteria/Nasienriglyne

- Must be a question; including the ?/Moet 'n vraag wees; ? ingesluit. ✓
- A relationship between correct variables must be given./'n Verwantskap tussen die korrekte veranderlikes moet gegee word. ✓

Example/Voorbeeld

What is the relationship between pressure and volume (of an enclosed gas)?

Wat is die verwantskap tussen druk en volume (van 'n ingeslote gas)? (2)

1.2 X: One of: ✓

Enclosed gas
Gas/Air
Glass tube

Een van: ✓

Ingeslote gas
Gas/Lug
Glasbuis

Y: One of: ✓

Bourdon pressure gauge
Pressure gauge

Een van: ✓

Bourdon drukmeter
Drukmeter

Z: One of: ✓

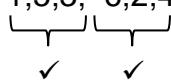
Oil container/reservoir
Oil

Een van: ✓

Oliehouer/-reservoir
Olie

(3)

1.3 1;5;3; 6;2;4



(2)

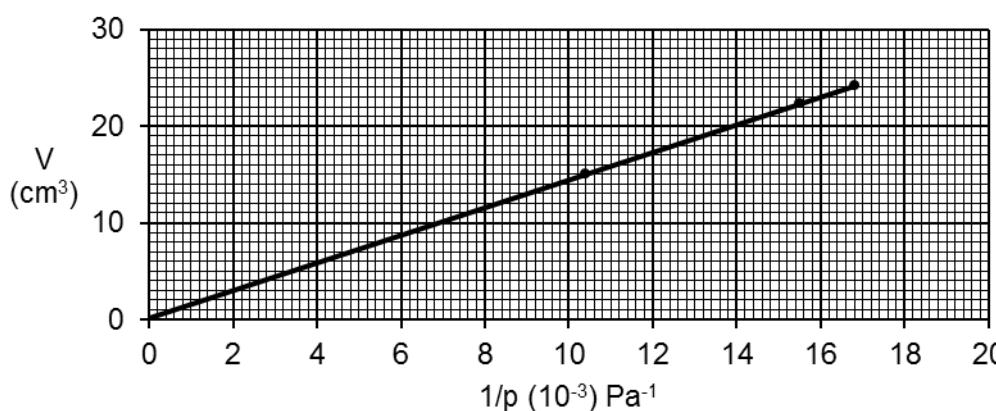
1.4 Increase ✓

Neem toe ✓

(1)

1.5.1

Graph of V versus 1/p
Grafiek van V teenoor 1/p



(3)

Marking criteria / Nasienriglyne

- Division (i.e. all numbers can fit and spacing is regular) of both axes is correct./Indeling (d.i. alle getalle pas in en spasiëring is reëlmatig) op beide asse is korrek. ✓
- Points are plotted correctly./Punte is korrek gestip. ✓ [This mark is forfeited if division mark is lost./Hierdie punt word verbeur as indelingspunt verloor is.]
- Best-fit straight line is drawn correctly from plotted points, even if mark for plotting is lost.
Bes-passende reguitlyn is korrek getrek vanaf gestippte punte, selfs as punt vir stip verloor is. ✓

- 1.5.2 S: 96,2 (96,15)
T: 14,0 (14 or/of 14,01)
Q: 1 438 (1438,35)

Marking criteria / Nasienriglyne

- Anyone of the numbers correct/*Enigeen van die getalle korrek ✓*
- 2nd and 3rd numbers also correct/*2e en 3e getalle ook korrek ✓*

(2)

Typing error: Q1.6 does not exist.

Tikfout: V1.6 bestaan nie.

- 1.6 Temperature/Temperatuur ✓

Accept: Number of moles of gas/mass of gas

Aanvaar: Aantal mol gas/massa gas

(1)

- 1.7 Pressure is inversely proportional to volume (at constant temperature). ✓
OR Volume is inversely proportional to pressure (at constant temperature).
OR $p \propto 1/p$ (at constant temperature)

Druk is omgekeerd eweredig aan volume (by konstante temperatuur).

OF Volume is omgekeerd eweredig aan druk (by konstante temperatuur).

OF $p \propto 1/p$ (by konstante temperatuur)

(1)

[15]